

ANTHOCEPHALUS ALKALOIDS: ISODIHYDROCADAMBEINE.

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A recent communication¹ described the isolation and structure determination of cadambine and 3 α -dihydrocadambine (2) from Anthocephalus cadamba. We now report a third glycosidic indole alkaloid - isodihydrocadambine - for which we propose the structure 3a containing a novel N-4 - C-19 bond.

After gel permeation, ion exchange and partition chromatography of a methanolic extract, isodihydrocadambine was isolated as the acetate derivative [α]_D²⁵ -66° (MeOH), for which a molecular formula of C₃₇H₄₄N₂O₁₅ was established by mass measurement. The molecular ion of the corresponding propionate derivative differed by 70 m. u., indicating five acylable groups; none involved nitrogen since all were deacylated by brief treatment with sodium methoxide, and the basic function must therefore be a tertiary amine. Mild catalytic hydrogenation had no effect, so no readily reducible double bonds were present.

The UV spectrum was consistent with the presence of indole and methyl β -alkoxyacrylate chromophores, which were confirmed by signals in the NMR spectrum at τ 2.25 (s, >NH), 2.4-3.2 (m, 4 aromatic H) 2.61 (s, H-17) and 6.42 (s, -CO₂CH₃) and in the IR at 3485 (>NH) 1710 and 1645 (-O-C=C-C=O) cm⁻¹. Extension of the structure to a tetrahydro- β -carboline was possible from mass measurement of the typical ions at m/e 184, 170, 169 and 156², and also by oxidation to a 3,4-dihydro- β -carbolinium chromophore (λ _{max} 372 nm) with mercuric acetate.

A further characteristic³ mass spectral series at m/e 381 (M-331) 365 (M-347), 331, 169 and 109 indicated a hexoside tetraacetate, which also had NMR signals at τ 4.7 - 5.3 (H-1', 2', 3', 4'), 5.82 (H₂-6') 6.4 (H-5') and 7.91 - 8.02 (4Ac). Deacetylation and cleavage with β -D-glucosidase established the identity of the sugar. Unlike the glucoside, on addition of alkali the UV spectrum of the aglucone underwent a rapid, large increase in absorption at ca. 275 nm attributable to ionization of a β -hydroxyacrylate chromophore. Since the mass spectrum indicated that the sugar was not attached as an enol ether the

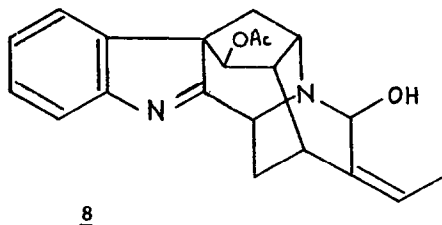
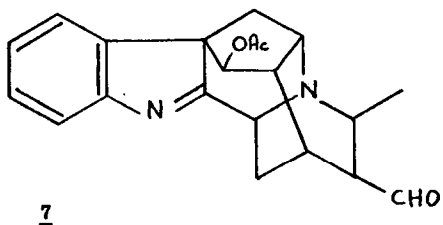
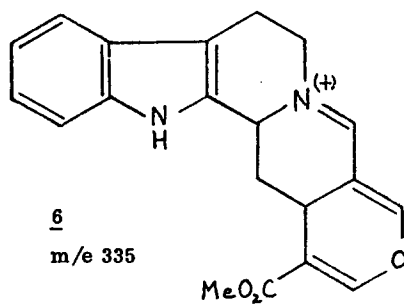
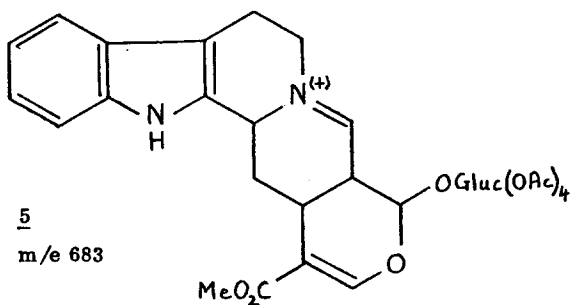
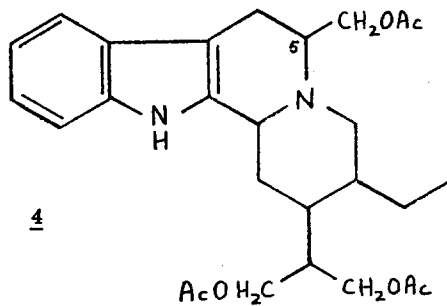
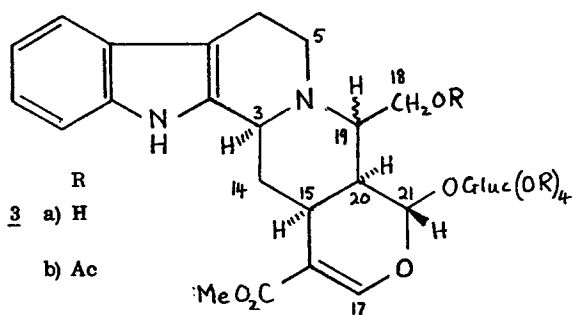
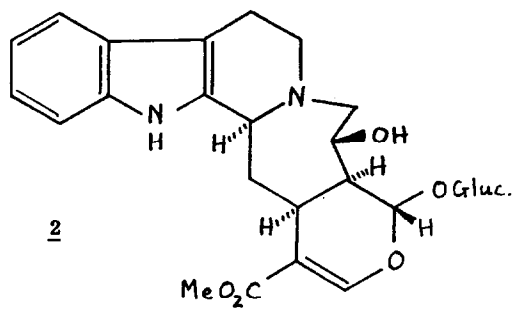
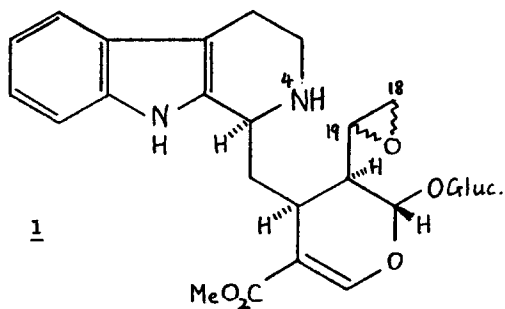
newly liberated hydroxyl group must form part of a labile hemi-acetal system: $\text{MeO}_2\text{C} - \overset{\overset{|}{\text{C}}}{\text{C}} = \overset{\overset{|}{\text{C}}}{\text{C}} - \overset{\overset{|}{\text{C}}}{\text{C}} - \text{OH}$.

The partial structure so far apparent was reminiscent of dihydrocadambine and derived from very similar spectral and chemical data. However, one striking difference was an intense ion at m/e 683 ($\text{C}_{34}\text{H}_{39}\text{N}_2\text{O}_{13}$) in the mass spectrum of isodihydrocadambine acetate which had no counterpart in the other. This was attributed to a loss of a $-\text{CH}_2\text{OAc}$ group so facile as to be consistent only with the cleavage of a C - C bond α to N-4 where the resulting ion was stabilised by the lone pair. A close analogy was provided by adirubine triacetate (4)⁴ whose base peak was formed by loss of $-\text{CH}_2\text{OAc}$ from C-5 which was likewise α to nitrogen. However, in the present case substitution at C-5 was excluded by the occurrence of the β -carbolinium fragments at m/e 184, 170, 169 rather than 182, 169, 168², and the dehydrogenation by mercuric acetate eliminated C-3 as a likely site. The $-\text{CH}_2\text{OAc}$ must therefore have been part of the third alkyl group of the tertiary amine outside ring C. Assuming a reasonable biogenesis from secologanin the structure 3b for the acetate was consistent with the above data. Furthermore, it readily explained the base peak at m/e 335 ($\text{C}_{20}\text{H}_{19}\text{N}_2\text{O}_3$) as due to 6 formed by a favourable retro-Michael elimination from the immonium ion 5 - a process supported by metastable ions at 616.6 and 164.3.

A detailed examination of the NMR spectrum with the aid of decoupling allowed most of the hydrogens to be assigned in accordance with structure 3b and also provided stereochemical information. Thus H-21 (τ 4.16) had a trans-diaxial interaction (9.5 Hz) with H-20 (τ 7.84), which had a coupling (\sim 5Hz, cis) with H-15 and also with H-19 (τ 7.04) although in the latter case the constant did not reveal the stereochemistry; in turn H-19 was coupled to the C-18 methylene group found as an AB pair at τ 5.5. Another ABX set of signals at τ 6.30 and 8.2 - 8.6 was attributed to H-3 and the C-14 methylene group respectively. By analogy with standard tetra- and pentacyclic indole alkaloids the chemical shift of H-3 suggested a 3α configuration (cis to H-15), which was supported by trans-quinolizidine IR bands in the 2700 - 2860 cm^{-1} region. Finally in the CD spectrum a strong positive Cotton effect between 300 and 260 nm established the absolute configuration at C-3 as α ⁵.

Hence isodihydrocadambine corresponds to 3a containing a N-4 - C-19 bond which is almost unique - the only other example being perakine (7), a probable artefact of vomemiline (8)⁶. It is presumably formed from an 18,19-epoxystrictosidine (1) by nucleophilic attack of N-4 on C-19 rather than C-18 as with dihydrocadambine (2). Whether isodihydrocadambine comes from the 19S epoxide like dihydrocadambine

or is derived from the 19R remains an open question.



References

1. R. T. Brown and S. B. Fraser, Tetrahedron Letters, 1957, (1974).
2. L. D. Antonaccio, N. A. Pereira, B. Gilbert, H. Vorbrueggen, H. Budzikiewicz, J. M. Wilson and C. Djerassi, J.A.C.S., 84, 2161 (1962).
3. R. T. Brown and L. R. Row, Chem. Comm., 453 (1967).
4. R. T. Brown, C. L. Chapple and G. K. Lee, Chem. Comm., 1007 (1972).
5. W. Klyne, R. J. Swan, N. J. Dastoor, A. A. Gorman and H. Schmid, Helv. Chim. Acta., 50, 115 (1960). C. M. Lee, W. T. Trager and A. H. Beckett, Tetrahedron, 23, 375 (1967).
6. W. I. Taylor, A. J. Frey and A. Hofmann, Helv. Chim. Acta., 45, 611 (1962).